

Class - VII

SUBJECT- CHEMISTRY

CHAPTER- Revision

TEACHER- ANAMIKA

Q:- Write the names of the following Compounds

a) $(\text{NH}_4)_2\text{SO}_4 \rightarrow$ Ammonium sulphate

b) $\text{Ca}(\text{NO}_3)_2 \rightarrow$ Calcium nitrate

c) $\text{FeS} \rightarrow$ Iron sulphide

d) $\text{Na}_3\text{PO}_4 \rightarrow$ Sodium phosphate

e) $\text{NH}_4\text{OH} \rightarrow$ Ammonium hydroxide

f) $\text{CuCO}_3 \rightarrow$ Copper carbonate

g) $\text{HgO} \rightarrow$ Mercuric oxide

h) $\text{ZnCl}_2 \rightarrow$ Zinc Chloride

i) $\text{ZnS} \rightarrow$ Zinc sulphide

j) $\text{H}_2\text{S} \rightarrow$ Hydrogen sulphide

Q:- What information do you get from the equation.



Ans:- When hydrogen and chlorine react to give hydrogen chloride.

Q:- (i) Name a metal used in making electric cable $\rightarrow$ Cu

(ii) Name a non metal used to kill germs in water - Cl_2

(iii) Name noble gas used for cancer therapy $\rightarrow$ Radon

(iv) Name non metal that is good conductor of heat & electricity $\rightarrow$ Carbon (Diamond)

(v) Name a noble gas used for filling into electric bulb $\rightarrow$ Argon.